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4-({4-[Bis(2-cyanoethyl)amino]phenyl}-
diazanyl)benzenesulfonamide

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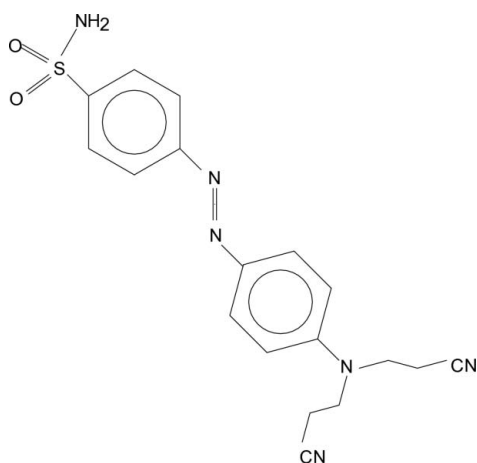
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Key indicators: single-crystal X-ray study; $T = 295$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.050; wR factor = 0.055; data-to-parameter ratio = 16.8.

In the title compound, $\text{C}_{18}\text{H}_{16}\text{N}_6\text{O}_2\text{S}$, which belongs to the family of azo dyes, the dihedral angle between the benzene rings is $26.16(7)^\circ$. In the crystal, molecules are joined by $\text{N}-\text{H}\cdots\text{N}$ and $\text{C}-\text{H}\cdots\text{N}$ hydrogen bonds into double chains parallel to the a axis.

Related literature

For the synthesis and properties of azo dyes, see: Wenker (1935); Ledoux *et al.* (2000); Viscardi *et al.* (2002). For a related structure, see: Sasaki *et al.* (2004).



Experimental

Crystal data

 $\text{C}_{18}\text{H}_{16}\text{N}_6\text{O}_2\text{S}$ $M_r = 382.45$

Triclinic, $P\bar{1}$
 $a = 7.8093(16)$ Å
 $b = 11.035(2)$ Å
 $c = 11.776(3)$ Å
 $\alpha = 94.268(4)^\circ$
 $\beta = 106.544(4)^\circ$
 $\gamma = 104.568(5)^\circ$

$V = 929.8(3)$ Å³
 $Z = 2$
 Mo $K\alpha$ radiation
 $\mu = 0.20$ mm⁻¹
 $T = 295$ K
 $0.30 \times 0.23 \times 0.03$ mm

Data collection

Siemens-Bruker APEX
 diffractometer
 Absorption correction: multi-scan
 (SORTAV; Blessing, 1995)
 $T_{\min} = 0.93$, $T_{\max} = 1.00$
 9271 measured reflections

4100 independent reflections
 1577 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.040$
 15 standard reflections every 60 min
 intensity decay: none

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.050$
 $wR(F^2) = 0.055$
 $S = 0.86$
 4100 reflections
 244 parameters

H atoms treated by a mixture of
 independent and constrained
 refinement
 $\Delta\rho_{\max} = 0.40$ e Å⁻³
 $\Delta\rho_{\min} = -0.48$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C15}-\text{H15A}\cdots\text{N23}^{\text{i}}$	0.93	2.52	3.427 (4)	166
$\text{N10}-\text{H10B}\cdots\text{N27}^{\text{ii}}$	0.91 (2)	2.19 (2)	3.084 (2)	165
$\text{N10}-\text{H10A}\cdots\text{N12}^{\text{iii}}$	0.94 (2)	2.19 (2)	3.124 (2)	176

Symmetry codes: (i) $-x + 2, -y, -z + 1$; (ii) $-x, -y, -z + 1$; (iii) $x - 1, y, z$.

Data collection: SMART (Bruker, 2007); cell refinement: SAINT (Bruker, 2007); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: XP in SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXL97.

We thank Dr C. Barolo for supplying crystals of the title compound.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: RZ2541).

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supplementary materials

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4-({4-[Bis(2-cyanoethyl)amino]phenyl}diazenyl)benzenesulfonamide

G. Gervasio, D. Marabello and F. Bertolotti

Comment

The blood gas (CO₂ or O₂) measurement is often required in modern diagnosis and the indicator properties of azo dyestuff have been proved very informative in this field, as proposed in the pioneering work of Wenker (1935). The title compound, 4-diethylcyanoamino-4'-sulfonylamino-azobenzene (Fig. 1), is part of this study and has been synthesized according to a modification of a standard procedure (Ledoux *et al.*, 2000; Viscardi *et al.*, 2002). Bond lengths and angles agree with those of a similar compound reported by Sasaki *et al.* (2004). The two phenyl rings form a dihedral angle of 26.16 (7)°. In the crystal packing, double chains of molecules parallel to the *a* axis are generated by weak C—H···N and N—H···N hydrogen bonds, where N acceptor atoms are the azo (N12) or cyano (N27) atoms (Table 1).

Experimental

The title compound has been obtained according to Ledoux *et al.* (2000) and Viscardi *et al.* (2002), and prepared by a modification of a standard procedure in analogy to similar compounds previously reported. Crystals suitable for X-ray analysis were obtained by slow evaporation of an ethanol solution.

Refinement

All H atoms, except those of the NH₂ group, have been placed in geometrically idealized positions and refined as riding, with C—H = 0.93–0.97 Å and with $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$. The NH₂ hydrogen atoms have been located in the final Fourier map and refined freely with $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{N})$. A small and poorly diffracting crystal has been used in the analysis.

Figures

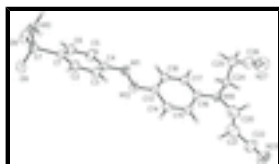


Fig. 1. The molecular structure of the title compound showing the atomic numbering and 50% probability displacements ellipsoids.

4-({4-[Bis(2-cyanoethyl)amino]phenyl}diazenyl)benzenesulfonamide

Crystal data

C₁₈H₁₈N₆O₂S

$M_r = 382.45$

Triclinic, $P\bar{1}$

Hall symbol: -P 1

$Z = 2$

$F(000) = 400$

$D_x = 1.359 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

supplementary materials

$a = 7.8093$ (16) Å
 $b = 11.035$ (2) Å
 $c = 11.776$ (3) Å
 $\alpha = 94.268$ (4)°
 $\beta = 106.544$ (4)°
 $\gamma = 104.568$ (5)°
 $V = 929.8$ (3) Å³

Cell parameters from 52 reflections
 $\theta = 1.9$ – 20.2 °
 $\mu = 0.20$ mm⁻¹
 $T = 295$ K
Plate, red
 $0.30 \times 0.23 \times 0.03$ mm

Data collection

Siemens–Bruker APEX
diffractometer

Radiation source: fine-focus sealed tube

graphite

φ scans

Absorption correction: multi-scan
(*SORTAV*; Blessing, 1995)

$T_{\min} = 0.93$, $T_{\max} = 1.00$

9271 measured reflections

4100 independent reflections

1577 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.040$

$\theta_{\max} = 28.4$ °, $\theta_{\min} = 1.8$ °

$h = -10 \rightarrow 10$

$k = -14 \rightarrow 14$

$l = -14 \rightarrow 15$

15 standard reflections every 60 min

intensity decay: none

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.050$

$wR(F^2) = 0.055$

$S = 0.86$

4100 reflections

244 parameters

0 restraints

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2)]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.40$ e Å⁻³

$\Delta\rho_{\min} = -0.48$ e Å⁻³

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	−0.4369 (3)	0.3324 (2)	0.1462 (2)	0.0329 (7)
C2	−0.3729 (3)	0.2269 (2)	0.1422 (2)	0.0385 (7)
H2A	−0.4475	0.1532	0.0896	0.046*
C3	−0.1985 (3)	0.2304 (2)	0.2160 (2)	0.0397 (7)
H3A	−0.1535	0.1603	0.2117	0.048*
C4	−0.0912 (3)	0.3396 (2)	0.2967 (2)	0.0338 (7)
C5	−0.1585 (3)	0.4427 (2)	0.3021 (2)	0.0474 (8)
H5A	−0.0880	0.5146	0.3582	0.057*
C6	−0.3308 (3)	0.4404 (2)	0.2245 (2)	0.0438 (8)
H6A	−0.3735	0.5118	0.2258	0.053*
S7	−0.66133 (10)	0.32583 (7)	0.05031 (7)	0.0441 (2)
O8	−0.6950 (2)	0.44536 (15)	0.07477 (14)	0.0586 (6)
O9	−0.6729 (2)	0.27865 (16)	−0.06983 (13)	0.0548 (6)
N10	−0.8133 (3)	0.2210 (2)	0.0862 (2)	0.0468 (7)
H10A	−0.817 (3)	0.243 (2)	0.1638 (18)	0.056*
H10B	−0.799 (3)	0.1427 (19)	0.0698 (19)	0.056*
N11	0.0845 (3)	0.34976 (19)	0.38119 (16)	0.0416 (6)
N12	0.1748 (3)	0.28313 (18)	0.34648 (16)	0.0402 (6)
C13	0.3437 (3)	0.2869 (2)	0.4342 (2)	0.0343 (7)
C14	0.4404 (3)	0.2039 (2)	0.4088 (2)	0.0406 (8)
H14A	0.3980	0.1541	0.3336	0.049*
C15	0.5974 (3)	0.1942 (2)	0.4928 (2)	0.0416 (8)
H15A	0.6601	0.1382	0.4737	0.050*
C16	0.6644 (3)	0.2679 (2)	0.6072 (2)	0.0373 (7)
C17	0.5725 (3)	0.3570 (2)	0.6288 (2)	0.0388 (7)
H17A	0.6186	0.4110	0.7019	0.047*
C18	0.4168 (3)	0.3660 (2)	0.5447 (2)	0.0365 (7)
H18A	0.3586	0.4259	0.5614	0.044*
N19	0.8150 (3)	0.2551 (2)	0.69548 (18)	0.0471 (7)
C20	0.9028 (4)	0.1477 (3)	0.6819 (2)	0.0713 (10)
H20A	0.9460	0.1199	0.7584	0.086*
H20B	0.8126	0.0760	0.6252	0.086*
C21	1.0583 (4)	0.1988 (3)	0.6383 (2)	0.0766 (10)
H21A	1.1448	0.2737	0.6924	0.092*
H21B	1.0145	0.2216	0.5594	0.092*
C22	1.1544 (4)	0.0919 (3)	0.6327 (3)	0.0738 (11)
N23	1.2304 (4)	0.0229 (2)	0.6272 (2)	0.0917 (10)
C24	0.8777 (3)	0.3258 (2)	0.8154 (2)	0.0445 (8)
H24A	0.8795	0.4133	0.8098	0.053*
H24B	1.0046	0.3248	0.8548	0.053*
C25	0.7577 (3)	0.2751 (2)	0.8932 (2)	0.0538 (8)
H25A	0.8088	0.3273	0.9717	0.065*
H25B	0.6327	0.2814	0.8571	0.065*
C26	0.7480 (4)	0.1423 (3)	0.9075 (3)	0.0547 (9)
N27	0.7442 (3)	0.0422 (2)	0.9183 (2)	0.0735 (9)

supplementary materials

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0227 (18)	0.0427 (19)	0.0314 (16)	0.0090 (15)	0.0055 (14)	0.0084 (14)
C2	0.0346 (19)	0.0360 (18)	0.0405 (17)	0.0106 (15)	0.0069 (15)	-0.0041 (14)
C3	0.040 (2)	0.041 (2)	0.0427 (17)	0.0215 (16)	0.0118 (15)	0.0065 (15)
C4	0.030 (2)	0.0396 (19)	0.0351 (17)	0.0149 (16)	0.0098 (14)	0.0099 (15)
C5	0.039 (2)	0.045 (2)	0.0466 (18)	0.0112 (16)	-0.0015 (16)	-0.0058 (15)
C6	0.038 (2)	0.041 (2)	0.0496 (18)	0.0204 (16)	0.0025 (16)	-0.0011 (16)
S7	0.0330 (5)	0.0526 (6)	0.0437 (5)	0.0171 (4)	0.0030 (4)	0.0080 (4)
O8	0.0472 (14)	0.0501 (14)	0.0769 (14)	0.0306 (11)	0.0030 (11)	0.0080 (11)
O9	0.0463 (14)	0.0874 (16)	0.0290 (11)	0.0259 (11)	0.0041 (10)	0.0059 (11)
N10	0.0328 (15)	0.0556 (18)	0.0497 (16)	0.0109 (14)	0.0127 (13)	0.0012 (15)
N11	0.0277 (15)	0.0549 (17)	0.0401 (14)	0.0191 (13)	0.0017 (12)	0.0056 (12)
N12	0.0299 (15)	0.0549 (17)	0.0363 (14)	0.0157 (12)	0.0074 (12)	0.0094 (12)
C13	0.0276 (19)	0.0417 (19)	0.0362 (17)	0.0148 (15)	0.0094 (15)	0.0078 (15)
C14	0.039 (2)	0.051 (2)	0.0302 (16)	0.0160 (16)	0.0076 (15)	-0.0014 (15)
C15	0.043 (2)	0.054 (2)	0.0379 (17)	0.0302 (17)	0.0143 (15)	0.0024 (16)
C16	0.0285 (19)	0.049 (2)	0.0380 (18)	0.0181 (15)	0.0089 (15)	0.0108 (15)
C17	0.0333 (19)	0.0476 (19)	0.0336 (17)	0.0197 (15)	0.0023 (14)	-0.0012 (14)
C18	0.0327 (19)	0.0416 (19)	0.0376 (17)	0.0182 (15)	0.0089 (15)	0.0028 (15)
N19	0.0448 (17)	0.0643 (18)	0.0404 (15)	0.0410 (15)	0.0056 (13)	0.0005 (13)
C20	0.049 (2)	0.107 (3)	0.044 (2)	0.001 (2)	0.0126 (18)	0.0098 (19)
C21	0.078 (3)	0.073 (3)	0.064 (2)	0.006 (2)	0.013 (2)	0.016 (2)
C22	0.078 (3)	0.079 (3)	0.080 (2)	0.052 (2)	0.024 (2)	0.008 (2)
N23	0.113 (3)	0.094 (2)	0.121 (2)	0.076 (2)	0.069 (2)	0.040 (2)
C24	0.043 (2)	0.059 (2)	0.0342 (17)	0.0288 (16)	0.0034 (15)	0.0058 (16)
C25	0.064 (2)	0.061 (2)	0.0455 (19)	0.0304 (18)	0.0190 (17)	0.0099 (17)
C26	0.051 (2)	0.061 (2)	0.054 (2)	0.015 (2)	0.0202 (17)	0.005 (2)
N27	0.069 (2)	0.055 (2)	0.095 (2)	0.0141 (18)	0.0277 (17)	0.0099 (18)

Geometric parameters (\AA , $^\circ$)

C1—C6	1.364 (3)	C15—C16	1.405 (3)
C1—C2	1.381 (3)	C15—H15A	0.9300
C1—S7	1.775 (2)	C16—N19	1.376 (3)
C2—C3	1.381 (3)	C16—C17	1.403 (3)
C2—H2A	0.9300	C17—C18	1.363 (3)
C3—C4	1.387 (3)	C17—H17A	0.9300
C3—H3A	0.9300	C18—H18A	0.9300
C4—C5	1.373 (3)	N19—C24	1.447 (3)
C4—N11	1.419 (3)	N19—C20	1.530 (3)
C5—C6	1.388 (3)	C20—C21	1.452 (3)
C5—H5A	0.9300	C20—H20A	0.9700
C6—H6A	0.9300	C20—H20B	0.9700
S7—O8	1.4326 (15)	C21—C22	1.556 (3)
S7—O9	1.4403 (15)	C21—H21A	0.9700
S7—N10	1.610 (2)	C21—H21B	0.9700

N10—H10A	0.939 (19)	C22—N23	1.085 (3)
N10—H10B	0.913 (19)	C24—C25	1.527 (3)
N11—N12	1.259 (2)	C24—H24A	0.9700
N12—C13	1.414 (3)	C24—H24B	0.9700
C13—C14	1.390 (3)	C25—C26	1.472 (3)
C13—C18	1.393 (3)	C25—H25A	0.9700
C14—C15	1.372 (3)	C25—H25B	0.9700
C14—H14A	0.9300	C26—N27	1.116 (3)
C6—C1—C2	120.7 (2)	N19—C16—C17	120.6 (2)
C6—C1—S7	119.93 (19)	N19—C16—C15	121.9 (2)
C2—C1—S7	119.4 (2)	C17—C16—C15	117.5 (2)
C1—C2—C3	120.2 (2)	C18—C17—C16	121.3 (2)
C1—C2—H2A	119.9	C18—C17—H17A	119.4
C3—C2—H2A	119.9	C16—C17—H17A	119.4
C2—C3—C4	119.3 (2)	C17—C18—C13	121.0 (2)
C2—C3—H3A	120.4	C17—C18—H18A	119.5
C4—C3—H3A	120.4	C13—C18—H18A	119.5
C5—C4—C3	120.0 (2)	C16—N19—C24	122.0 (2)
C5—C4—N11	116.5 (2)	C16—N19—C20	122.3 (2)
C3—C4—N11	123.5 (2)	C24—N19—C20	114.63 (19)
C4—C5—C6	120.5 (2)	C21—C20—N19	106.4 (3)
C4—C5—H5A	119.7	C21—C20—H20A	110.4
C6—C5—H5A	119.7	N19—C20—H20A	110.4
C1—C6—C5	119.3 (2)	C21—C20—H20B	110.4
C1—C6—H6A	120.3	N19—C20—H20B	110.4
C5—C6—H6A	120.3	H20A—C20—H20B	108.6
O8—S7—O9	119.73 (10)	C20—C21—C22	106.0 (3)
O8—S7—N10	106.68 (11)	C20—C21—H21A	110.5
O9—S7—N10	106.08 (12)	C22—C21—H21A	110.5
O8—S7—C1	108.04 (11)	C20—C21—H21B	110.5
O9—S7—C1	107.70 (11)	C22—C21—H21B	110.5
N10—S7—C1	108.15 (12)	H21A—C21—H21B	108.7
S7—N10—H10A	112.7 (14)	N23—C22—C21	175.3 (4)
S7—N10—H10B	109.6 (15)	N19—C24—C25	114.2 (2)
H10A—N10—H10B	116 (2)	N19—C24—H24A	108.7
N12—N11—C4	114.1 (2)	C25—C24—H24A	108.7
N11—N12—C13	113.8 (2)	N19—C24—H24B	108.7
C14—C13—C18	118.2 (2)	C25—C24—H24B	108.7
C14—C13—N12	117.5 (2)	H24A—C24—H24B	107.6
C18—C13—N12	124.3 (2)	C26—C25—C24	112.5 (2)
C15—C14—C13	121.2 (2)	C26—C25—H25A	109.1
C15—C14—H14A	119.4	C24—C25—H25A	109.1
C13—C14—H14A	119.4	C26—C25—H25B	109.1
C14—C15—C16	120.6 (2)	C24—C25—H25B	109.1
C14—C15—H15A	119.7	H25A—C25—H25B	107.8
C16—C15—H15A	119.7	N27—C26—C25	178.6 (3)

supplementary materials

Hydrogen-bond geometry (Å, °)

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
C15—H15A···N23 ⁱ	0.93	2.52	3.427 (4)	166
N10—H10B···N27 ⁱⁱ	0.91 (2)	2.19 (2)	3.084 (2)	165
N10—H10A···N12 ⁱⁱⁱ	0.94 (2)	2.19 (2)	3.124 (2)	176

Symmetry codes: (i) $-x+2, -y, -z+1$; (ii) $-x, -y, -z+1$; (iii) $x-1, y, z$.

Fig. 1

